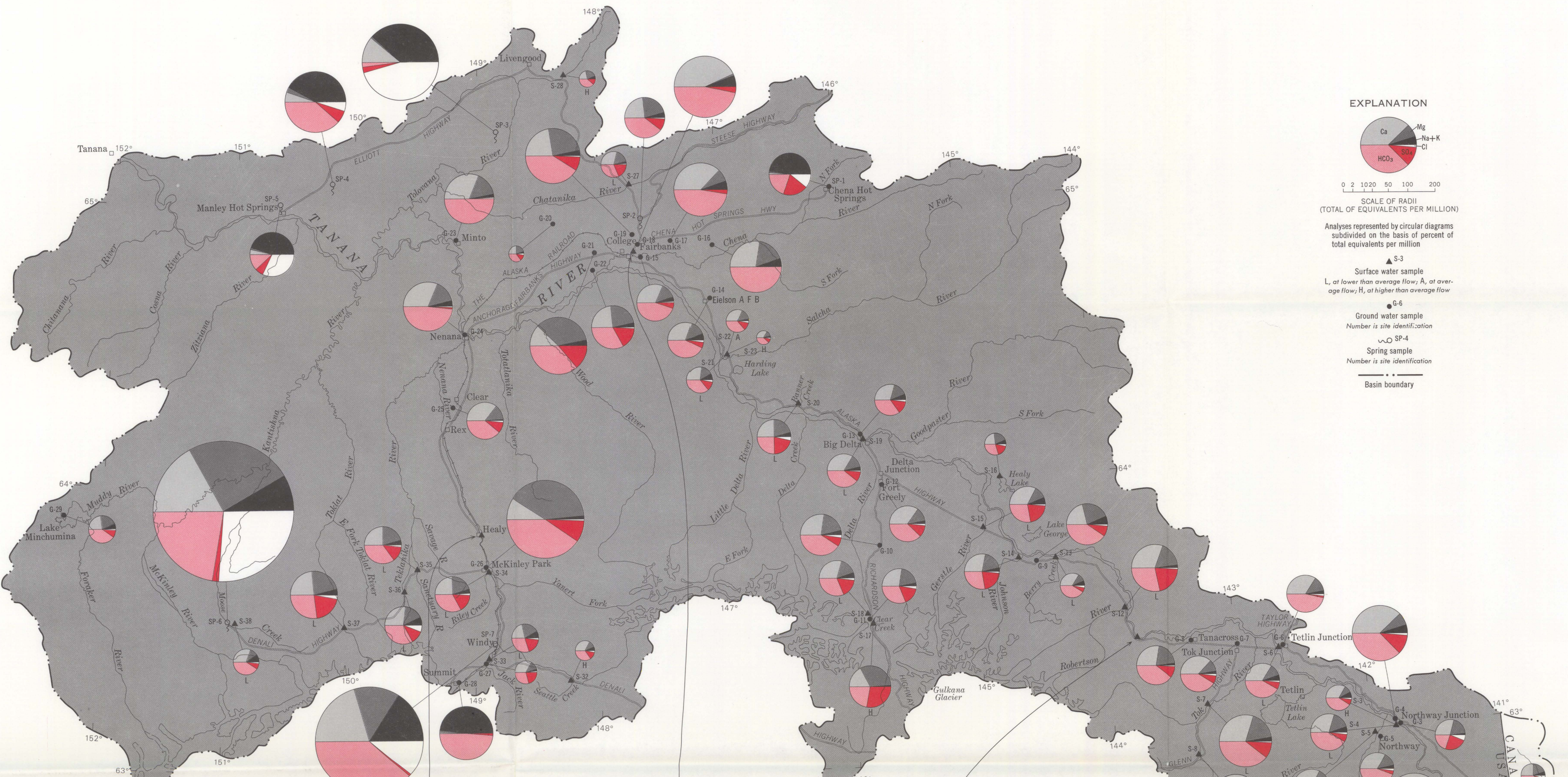
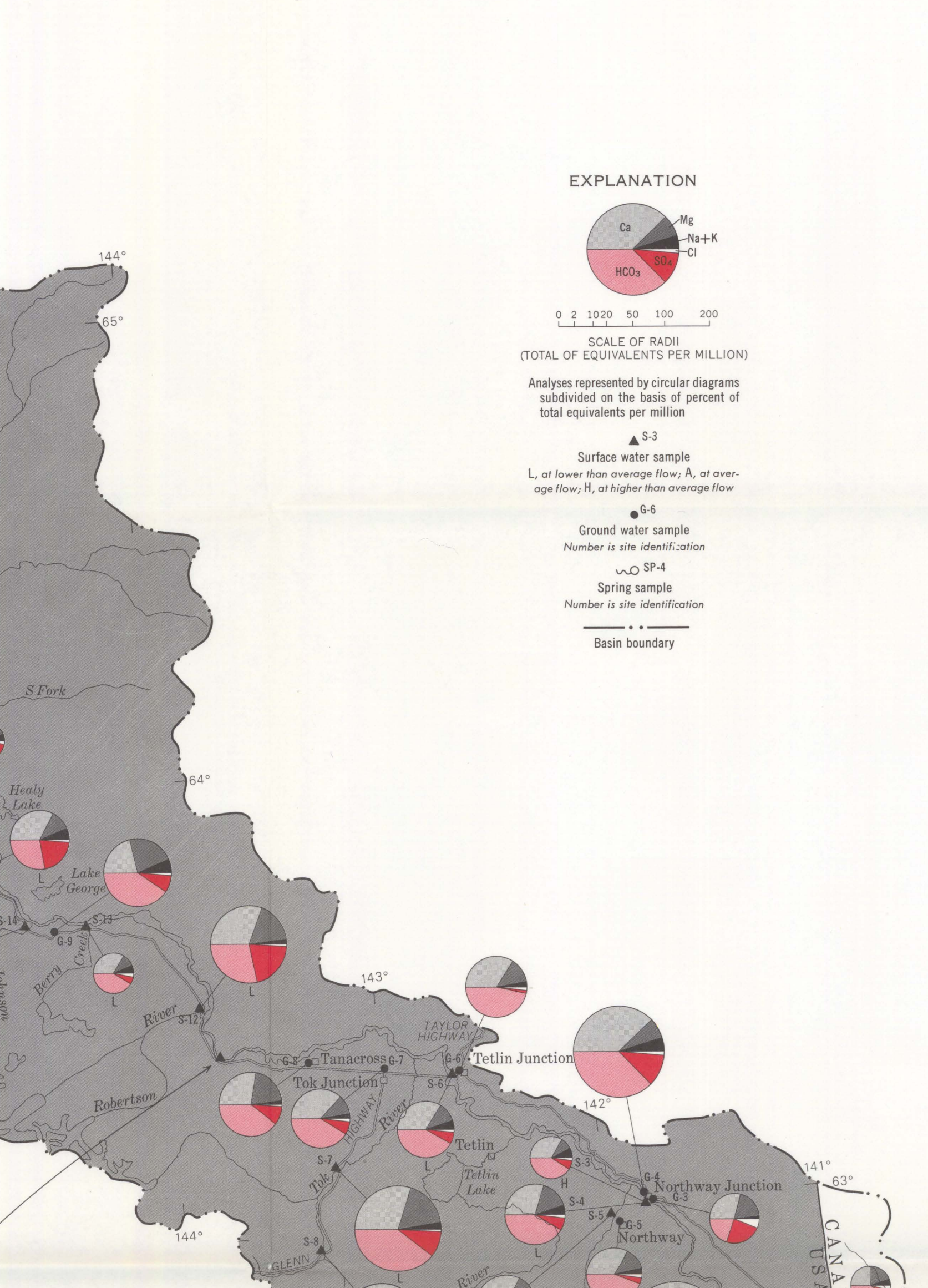
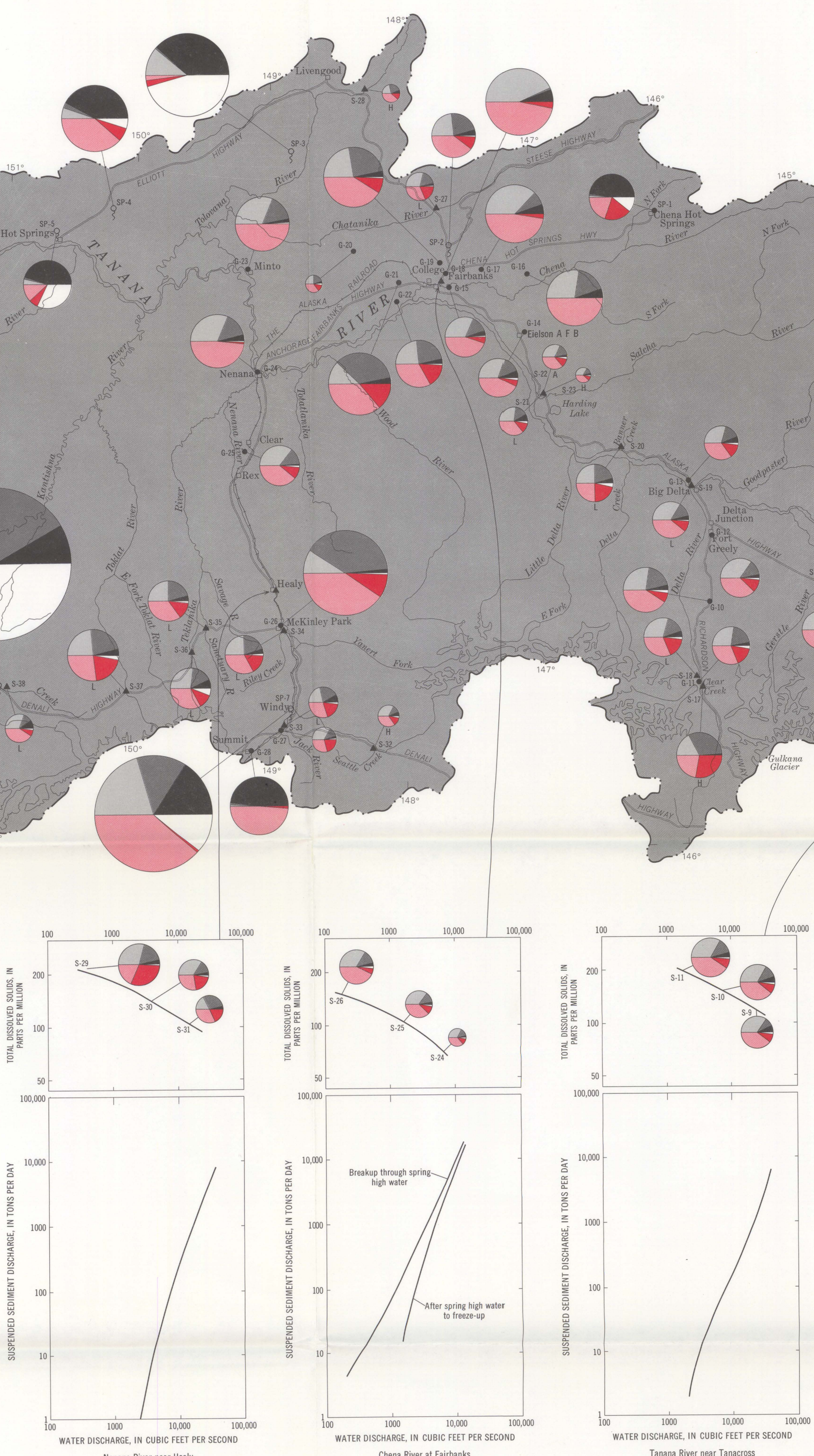


QUALITY OF WATER



Base from U.S. Geological Survey  
Fairbanks 1:250,000, 1933.  
Big Delta 1:250,000, 1958.  
Healy 1:250,000, 1956, and Mt.  
Haves 1:250,000, 1955



**CHEMICAL COMPOSITION**

The chemical composition of water in the Tanana basin is discussed under two headings, surface water, including streams and lakes, and ground water, including wells and springs. The adjacent map shows graphically the chemical types of water that are found in the basin. Lines on the map represent areas of similar chemical composition. During periods of low flow, the highest dissolved constituents are in streams draining mineralized bedrock areas, notably in the Alaska Range.

Chemical analyses of river water indicate that the higher flows have lower dissolved-solids concentration as shown by the graphs of the Chena, Nenana, and Tanana Rivers. This is because the peak discharges are derived from rapid runoff of rain or snowmelt which is low in dissolved mineral matter, whereas the low flow has a high proportion of ground-water inflow.

In general, the streams flowing from the Alaska Range are higher in sulfate and magnesium content than other streams in the basin. None of the streams that have been sampled, however, exceed standards suggested by the U.S. Public Health (1962) with respect to these constituents. The only constituent which does exceed the standards is iron. Iron in excess of the recommended limits has been found in two places, both near the Canadian border. One of these two waters is from a lake (S-18) analysis; the other is from a small stream (S-20) analysis. Both waters are shallow ground-water sources and the type of water discharged from these springs is similar to most ground water in the basin.

Most of the wells drilled along the boundaries of the basin tap igneous and metamorphic rocks and yield water that is high in magnesium bicarbonate or magnesium sulfate. Few of the wells that tap crystalline rocks contain more magnesium than the recommended limits. There, wells that tap the alluvial deposits and the streams that drain the bedrock areas also reflect the high magnesium sulfate. The highest nitrate concentration in the bedrock wells is generally, but not invariably, low in iron.

Wells drilled in the alluvial fans along the southern margin of the valley produce water that is relatively low in iron, moderate hardness, and of quality similar to the streams that flow across the alluvial fans. The alluvial fans provide the largest source of good-quality water in the Tanana basin.

Water from wells drilled in organic-rich sediments of the flood plains, terraces, and valley fill are low in sulfate, moderate to high in hardness, and moderate to high in iron. The origin of this water in the Fairbanks area is described by Colstrom (1962, p. 47); presumably his explanation applies to much of the Tanana basin lowland. As water passes through the organic-rich sediments, the sulfate is reduced to hydrogen sulfide and hydrogen sulfide, carbonic acid, formed from the carbon dioxide, attacks lime materials and increases hardness. Excessive carbon dioxide imparts corrosive properties to the water bringing iron into solution. The hydrogen sulfide imparts disagreeable odors to the water.

Most water in the Tanana basin can be characterized as calcium magnesium bicarbonate, commonly containing objectionable concentrations of iron. The quality of this water is improved by very poor but with municipal water requiring treatment for iron removal prior to use. Chloride and fluoride concentrations are low in all samples.

**GROUND WATER**

The chemical quality of ground water reflects its geologic environment. High springs in the area, such as S-1, are assumed to be connected with deposited sources that may account for high concentrations of sodium, chloride, bicarbonate, and magnesium. A few of the springs in the area, such as S-2, have shallow ground-water sources and the type of water discharged from these springs is similar to most ground water in the basin.

Most of the wells drilled along the boundaries of the basin tap igneous and metamorphic rocks and yield water that is high in magnesium bicarbonate or magnesium sulfate. Few of the wells that tap crystalline rocks contain more magnesium than the recommended limits. There, wells that tap the alluvial deposits and the streams that drain the bedrock areas also reflect the high magnesium sulfate. The highest nitrate concentration in the bedrock wells is generally, but not invariably, low in iron.

Wells drilled in the alluvial fans along the southern margin of the valley produce water that is relatively low in iron, moderate hardness, and of quality similar to the streams that flow across the alluvial fans. The alluvial fans provide the largest source of good-quality water in the Tanana basin.

Water from wells drilled in organic-rich sediments of the flood plains, terraces, and valley fill are low in sulfate, moderate to high in hardness, and moderate to high in iron. The origin of this water in the Fairbanks area is described by Colstrom (1962, p. 47); presumably his explanation applies to much of the Tanana basin lowland. As water passes through the organic-rich sediments, the sulfate is reduced to hydrogen sulfide and hydrogen sulfide, carbonic acid, formed from the carbon dioxide, attacks lime materials and increases hardness. Excessive carbon dioxide imparts corrosive properties to the water bringing iron into solution. The hydrogen sulfide imparts disagreeable odors to the water.

Most water in the Tanana basin can be characterized as calcium magnesium bicarbonate, commonly containing objectionable concentrations of iron. The quality of this water is improved by very poor but with municipal water requiring treatment for iron removal prior to use. Chloride and fluoride concentrations are low in all samples.

**SEDIMENT LOADS**

Sediment loads transported by streams have a direct effect on the cost and feasibility of water-resources development. High sediment loads have to be considered in reservoir design. Treatment is necessary if the water is to be used for domestic supply. Sediment deposition in streams reduces channel capacity and increases the flood potential. A comparison of the annual sediment yields of the three streams that have been sampled daily is shown below.

**ANNUAL SEDIMENT YIELDS IN THE TANANA BASIN**

Stream and location	Length of reach (years)	Drainage area (square miles)	Annual sediment yield		
			Tons per square mile	Min	
Tanana River near Tanacross	10	8,550	1,000	1,410	750
Chena River at Fairbanks	4	1,980	106	188	36
Nenana River near Healy	13	1,910	1,700	2,600	760

**SELECTED CHEMICAL ANALYSES OF SURFACE WATER**  
Chemical analyses in parts per million except carbonate, pH, and color

Number	Date of collection	Location	Mean discharge (cfs)	Silica (SiO <sub>2</sub> )	Iron (Fe)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Chloride (Cl)	Fluoride (F)	Nitrate (NO <sub>3</sub> )	Dissolved solids (residue on evaporation at 180°C)	Hardness as CaCO <sub>3</sub>	Specific conductance (micro-mhos at 25°C)	pH	Color
S-1	6/18/65	Lake Ina near Northway Junction	4.3	2.3	36	14	5.2	2.1	181	4.3	2.8	0.4	1.0	160	149	1	275	7.5	120
S-2	6/17/65	Scottie Creek near Northway Junction	6.2	0.41	15	6.4	3.2	1.3	70	8.6	2.1	0.3	0.5	78	64	7	134	7.3	120
S-3	7/27/57	Chena River near Northway Junction	7.1	0.06	26	12	0.6	0.8	98	13	2.0	0.1	0.3	102	85	11	185	8.0	0

**SELECTED CHEMICAL ANALYSES OF GROUND WATER**  
Chemical analyses in parts per million except carbonate, pH, and color

Number	Date of collection	Owner or user	Major aquifer	Depth of well (feet)	Silica (SiO <sub>2</sub> )	Iron (Fe)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Chloride (Cl)	Fluoride (F)	Nitrate (NO <sub>3</sub> )	Dissolved solids (residue on evaporation at 180°C)	Hardness as CaCO <sub>3</sub>	Specific conductance (micro-mhos at 25°C)	pH	Color	
																					Carbonate
G-1	3/10/66	Custom Station, Interat. Border	schist gravely-sand	297	9.0	0.02	37	335	66	1.8	494	1120	8.9	0.1	0.0	1820	1560	1185	2440	7.6	140
G-2	6/9/65	Border Trading Post	granite	175	26	2.28	135	25	17	7.9	602	5.3	2.8	0.4	0.3	329	438	0	899	7.6	140
G-3	11/2/59	Northway Model	granite	29	71	12	12	8.2	0.4	106	56	10	0.6	0.4	24	90	26	283	7.4	—	

**SELECTED CHEMICAL ANALYSES OF SPRINGS**  
Chemical analyses in parts per million except carbonate, pH, and color

Number	Date of collection	Location	Mean discharge (gpm)	Temperature (F)	Silica (SiO <sub>2</sub> )	Iron (Fe)	Calcium (Ca)	Magnesium (Mg)	Sodium (Na)	Potassium (K)	Bicarbonate (HCO <sub>3</sub> )	Sulfate (SO <sub>4</sub> )	Chloride (Cl)	Fluoride (F)	Nitrate (NO <sub>3</sub> )	Dissolved solids (residue on evaporation at 180°C)	Hardness as CaCO <sub>3</sub>	Specific conductance (micro-mhos at 25°C)	pH	Color	
																					Carbonate
SP-1	8/5/15	Chena Hot Springs	220	149	77	—	2.3	1.2	94	1.7	118	78	26	0.2	0.5	388	196	32	543	7.1	0
SP-2	9/30/66	Fox Spring	8	36	11	0.00	40	23	1.5	2.0	40	615	0.2	0.2	1180	210	170	2090	7.7	0	
SP-3	1/2/61	Tanana Hot Springs	124	75	10.2	1.2	321	1.2	49	40	615	0.2	0.2	1180	210	170	2090	7.7	0		

HYDROLOGIC RECONNAISSANCE OF THE TANANA RIVER, CENTRAL ALASKA

By  
G. S. Anderson  
1970